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Recent Advances in Borate-Based Glass Nanocomposites: Structural, Dielectric, and Radiation Shielding Properties

Taha Abdel Mohaymen Taha^{1*}, Ahmed El-Seady², S. El-Rabaie¹, A.A. El-Adawy²

¹Physics and Engineering Mathematics Department, Faculty of Electronic Engineering, Menoufia University, Menouf 32952, Egypt

²Physics Department, Faculty of Science, Menoufia University, Shebin El-Koom 32511, Egypt

*Correspondence Email: taha.hemida@yahoo.com

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ABSTRACT

Received: 07/08/2025 Accepted: 20/08/2025 Borate-based glass nanocomposites have gained significant attention in recent years due to their unique structural, dielectric, and radiation shielding properties. These materials exhibit excellent thermal stability, tunable optical characteristics, and enhanced mechanical strength, making them suitable for applications in optoelectronics, solid-state batteries, and radiation protection. This comprehensive review provides an in-depth analysis of recent advancements in borate-based glass nanocomposites, with a particular focus on three key aspects: (1) structural modifications induced by different compositional changes, (2) dielectric behavior and their dependence on glass network formers and modifiers, and (3) their effectiveness as radiation shielding materials. The influence of dopants, such as transition metals (e.g., Cu, Fe, Co) and rare-earth elements (e.g., Ce, Eu, Gd), on the optical, electrical, and mechanical properties of these glasses is systematically discussed. These dopants play a crucial role in modifying the glass network, enhancing ionic conductivity, and improving radiation attenuation performance. The influence of dopants, such as transition metals and rare-earth elements, on the physical and chemical properties of these glasses is discussed. Additionally, the role of nanostructuring in improving radiation attenuation is explored, along with future prospects for these advanced materials.

Keywords: Borate glass, Glass-ceramic, Nanocomposite, Dielectric properties, Radiation shielding

1. Introduction

1.1. Overview of Borate Glasses

Borate (B₂O₃)-based glasses have been extensively studied due to their unique structural flexibility, low melting point, and high transparency in the visible and near-infrared regions (Kamitsos, 2025; Shilpa et al, 2025; Eskalen et al, 2025). Unlike silicate and phosphate glasses, borate glasses exhibit a distinctive

network structure composed of BO₃ trigonal units and BO₄ tetrahedral units, which can be modified by the addition of various oxides and dopants. This structural adaptability allows for the tailoring of physical, optical, and electrical properties, making borate glasses suitable for a wide range of technological applications, including optical fibers, laser hosts, and solid-state electrolytes (Möncke, 2015 et al; Topper & Möncke, 2022; Yiannopoulos et al 2001; Alderman, 2018).

1.4. Scope of This Review

1.2. Emergence of Borate-Based Glass Nanocomposites

Recent advancements in materials science have led to of development borate-based nanocomposites, where the incorporation of nanoparticles (metallic, ceramic, or rare-earth-doped) into the glass matrix enhances its functional properties. The addition of nanoscale fillers, such as ZnO (Eskalen et al, 2023), Fe₂O₃ (Ivanova et al, 2016), CeO₂ (Ranasinghe et al. 2019), or metallic nanoparticles (Ag (Ande et al, 2021), Au (Jagannath et al, 2018)), introduces novel characteristics, including improved mechanical strength, optical, and superior radiation shielding capabilities. These nanocomposites bridge the gap between traditional glasses and advanced functional materials, offering new possibilities for optoelectronics, energy storage, and radiation protection.

1.3. Importance of Structural, Dielectric, and Radiation Shielding Properties

The structural properties of borate glass nanocomposites are crucial in determining their thermal stability, chemical durability, and optical behavior. By carefully selecting dopants and controlling nanoparticle dispersion, researchers can engineer glasses with tailored properties for specific applications (Deepika et al, 2021).

Dielectric properties are particularly important for electronic and energy storage applications. Borate glass nanocomposites exhibit tunable permittivity and low dielectric loss, making them promising candidates for capacitors, insulators, and solid-state battery components. The presence of nanoparticles can further enhance interfacial polarization, leading to improved dielectric performance (Liu et al, 2019; Chakrabarti et al, 2022; Fan et al, 2018).

Radiation shielding is another critical application, especially in medical imaging, nuclear reactors and space technology (Mortazavi et al, 2024). Traditional shielding materials like lead and concrete have limitations in terms of weight, toxicity, and environmental impact (AbuAlRoos et al, 2019). Borate-based glass nanocomposites, especially those doped with high-Z elements (e.g., Bi, Ba, Gd), offer lightweight, transparent, and efficient alternatives for gamma-ray and neutron attenuation (Alhakami et al, 2025; Al-Buriahi et al, 2020; Intachai et al, 2020).

This review presents a comprehensive analysis of cutting-edge developments in borate-based glass nanocomposites, systematically examining how strategic structural modifications through dopants (alkali/alkaline earth/rare earth elements) and nanoparticle integration fundamentally alter the BO₃/BO₄ network configuration and interface dynamics, enabling unprecedented control over material properties. It explores breakthroughs in dielectric behavior optimization, where interfacial polarization effects from nanofillers (TiO₂, BaTiO₃) and ionic conduction pathways in Li-doped systems achieve exceptional permittivity (ε'>100) with minimal loss (tan δ <0.01), revolutionizing capacitor and solid-state battery technologies. The analysis highlights transformative radiation shielding performance. demonstrating high-Z how nanocomposites $(Bi_2O_3/Gd_2O_3-B_2O_3)$ surpass conventional materials in gamma/neutron attenuation offering transparency and lightweight advantages, supported by Monte Carlo simulations validating their superior mass attenuation coefficients. Addressing critical gaps, the review identifies key future challenges in scalable manufacturing (3D printing, AI-optimized eco-friendly alternatives (Pb/Cd compositions), replacement), and multifunctional integration (luminescent-sensing shields), while projecting their industrial potential in nuclear, aerospace, and medical sectors through advanced glass-ceramic hybrids and smart adaptive designs. By synthesizing these interdisciplinary advances, the review positions borate glass nanocomposites as versatile nextgeneration materials capable of meeting extreme technological demands through nanoscale-engineered structures and tunable functionalities.

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2. Structural Properties of Borate-Based Glass Nanocomposites

Borate glasses possess a unique structural framework that distinguishes them from other oxide glasses. The fundamental building blocks of these glasses are triangular BO3 units and tetrahedral BO4 units, which form a highly adaptable network structure. This flexibility allows for structural extensive modifications through the incorporation of various modifiers and dopants, making borate glasses particularly attractive for nanocomposite development. Recent advances in characterization techniques and computational modeling have provided deeper insights into the structural evolution of these materials at atomic, nano-, and macro-scales

(Edwards et al, 2014; Stone et al, 2000; Parthé, 2002; Parthé, 2004).

2.1. Fundamental Structural Units and Network Formation

The base B₂O₃ glass features a distinctive structure primarily composed of boroxol rings (B2O3) formed by three interconnected BO3 triangular units linked through bridging oxygen atoms, which creates an open network with low packing density while exhibiting exceptional glass-forming ability and the characteristic "boron anomaly" - a unique phenomenon where physical properties such as density, thermal expansion coefficient, and refractive index vary non-monotonically with composition changes due to the complex interplay between BO3 trigonal and BO₄ tetrahedral coordination states. This structural peculiarity arises from the ability of boron to readily change its coordination number in response to modifier additions, making the borate network particularly sensitive to compositional variations and enabling precise tuning of material properties through controlled doping or nanostructuring (Fig. 1).

The introduction of network modifiers such as alkali (e.g., Li₂O, Na₂O) or alkaline earth oxides (e.g., CaO, MgO) induces significant structural transformations in borate glasses, initially converting planar BO₃ units into tetrahedral BO4 units up to a critical modifier concentration (typically around 30-40 mol%), which enhances network connectivity and modifies physical properties, while at higher modifier contents the excess oxygen leads to the formation of non-bridging oxygens (NBOs) that disrupt the network and create more open structures, with these coordination changes (BO₃→BO₄→NBO formation) causing non-linear variations in key physical properties like density, thermal expansion, and mechanical strength, known collectively as the "boron anomaly" that characterizes the unique structure-property relationships in borate glass systems. These structural modifications form the basis for tailoring borate glasses for specific applications through controlled compositional adjustments (Padmaja & Kistaiah, 2009; Kojima, 2020, Lower et al, 2001; Button et al, 1982; Shashikala & Udayashankar, 2016).

Recent studies have demonstrated that strategic doping represents a powerful approach for precisely controlling the borate network structure at multiple length scales, enabling tailored material properties for specific technological applications. Advanced characterization techniques combined with computational modeling have revealed that different

dopant categories induce distinct structural modifications: alkali and alkaline earth ions (e.g., Li⁺, Ca^{2+}) primarily act as network modifiers. systematically converting BO₃ triangles to BO₄ tetrahedra up to characteristic concentrations while simultaneously affecting the overall network connectivity through charge compensation mechanisms (Kamitsos et al, 2020; Pimentel et al, 2018).

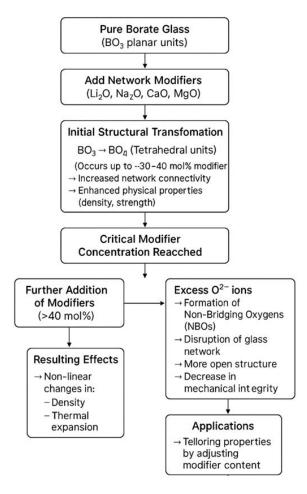


Figure 1: Structural evolution of borate glasses with network modifiers.

Transition metal dopants (e.g., Fe³⁺, Cu²⁺) exhibit more complex behavior, functioning as both network modifiers and conditional glass formers, with their multivalent nature introducing additional redox-dependent structural configurations that can be exploited for optoelectronic applications (Wright et al, 2017; Funabiki et al, 2011). Particularly significant are rare earth dopants (e.g., Gd³⁺, Eu³⁺) which, despite their large ionic radii, can be incorporated into the borate network through the formation of modified coordination polyhedra that create unique local environments while maintaining

glass homogeneity (Wantana et al, 2023; Fayaz et al, 2023: Marimuthu et al. 2009). Recent breakthroughs spectroscopy and high-resolution microscopy have shown that these dopant-induced structural changes occur not just at the atomic scale but propagate through medium-range order (2-5 Å) to influence the overall glass network topology. Furthermore, the synergistic effects of mixed dopants have been found to create novel structural motifs that cannot be achieved with single-component doping, opening new possibilities for property engineering. This precise structural control through strategic doping has become particularly valuable in developing advanced glass nanocomposites where the dopants not only modify the base glass structure but also influence nanoparticle-glass matrix interactions, interfacial bonding, and overall composite morphology. The ability to predict and control these structural modifications through machine learning approaches and ab initio calculations represent a significant advancement in glass science, moving from empirical composition-property relationships to rationally designed materials with predetermined characteristics (Rada et al, 2012; Ohkubo, 2016; Kato et al, 2024; Urata et al, 2024; Ahmmad et al; 2022).

2.2. Nanostructural Engineering and Characterization

The integration of nanoscale constituents (0D nanoparticles, 1D nanowires/nanotubes, or 2D nanosheets) into borate glass matrices creates complex interfacial regions that profoundly influence bulk material properties through multiple synergistic mechanisms. For metallic nanoparticles (Ag, Au) (Adamiv et al, 2014; Gurushantha et al 2023), the dielectric mismatch between the nanoparticles and glass host generates localized surface plasmon resonance effects, which not only produce vivid plasmonic colors but also enhance nonlinear optical properties and enable surface-enhanced Raman applications (see Fig. scattering 2). Oxide nanoparticles (TiO2 (Colak, 2017), ZrO2 (Shearer et al, 2024)) exhibit markedly different behavior, serving as heterogeneous nucleation sites that promote controlled crystallization during thermal processing while simultaneously reinforcing the glass network through a "filler effect" that improves mechanical strength and thermal stability.

Semiconductor quantum dots (CdSe, PbS) introduce quantum confinement effects that enable precise tuning of optoelectronic properties through sizedependent bandgap engineering, making them particularly valuable for luminescent and photovoltaic applications (Hayes et al, 2024; Fan et al, 2014).

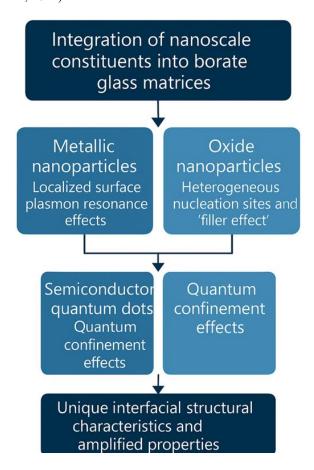


Figure 2: The integration of various nanoscale constituents into borate glass matrices.

The interfacial regions surrounding these nanoscale constituents exhibit unique structural characteristics typically featuring modified bond angles, altered coordination environments, and strain fields that extend several atomic layers into the glass matrix. These interfacial zones act as "property amplifiers," where small additions of nanomaterials (often <5 vol%) can induce disproportionate improvements in target properties due to the large surface-to-volume ratios and the resulting dominance of interfacemediated phenomena. Recent studies using advanced electron microscopy and spectroscopy techniques have revealed that these interfaces often exhibit graded compositions and distinct short-range ordering that differ substantially from both the nanoparticle cores and bulk glass matrix, creating what are effectively "third phases" with unique characteristics (Moriceau et al, 2019; Svecova et al, 2019). The ability to control nanoparticle dispersion, interface chemistry, and spatial distribution through advanced processing techniques (e.g., laser melting

(Seyfarth, et al, 2018), sol-gel methods (Lepry & Nazhat, 2015), or field-assisted sintering (Moriceau et al, 2019)) has opened new frontiers in designing borate glass nanocomposites with precisely tailored optical, mechanical, and functional properties for next-generation applications (Vijayakumar et al, 2018; Fonseca, 2022).

2.3. Advanced Characterization Techniques

characterization Modern techniques have revolutionized our understanding of borate glass nanocomposites by providing unprecedented insights into their structural hierarchy across multiple length High-resolution transmission electron scales. microscopy (HR-TEM), particularly when coupled with energy-dispersive X-ray spectroscopy (EDS) and electron energy loss spectroscopy (EELS), visualization of nanoparticle enables direct distribution, crystallographic orientation relationships, and the atomic-scale structure of glassnano interfaces, revealing critical details about interface bonding, strain fields, and potential reaction layers (Zhao et al, 2023). Synchrotron X-ray diffraction (XRD) techniques leverage the high brilliance and tunable energy of synchrotron sources to probe both the short-range (1-5 Å) and mediumrange (5-20 Å) order in these materials, providing pair distribution function (PDF) analysis that can distinguish between different borate structural units and their spatial correlations (Hannon, 2015). Solidnuclear resonance magnetic spectroscopy, especially 11B magic-angle spinning (MAS) NMR, serves as a quantitative tool for determining the BO₃/BO₄ ratio and characterizing the local coordination environments around boron atoms, while multinuclear NMR (²³Na, ²⁷Al, etc.) can track modifier ion distributions and their interactions with the glass network (Wang & Stebbins, 1999; Eckert, 2018) (see Fig. 3).

spectroscopy offers complementary Raman information by identifying specific borate structural units (boroxol rings, pentaborate groups, etc.) through their characteristic vibrational fingerprints, while also monitoring dopant-induced modifications through changes in peak positions, intensities, and bandwidths (Massot et al, 1995; Yano et al, 2003). When these techniques are combined with advanced data analysis methods (multivariate analysis, machine learning-assisted pattern recognition) and in-situ capabilities (high-temperature, pressure, or radiation environments), they provide a comprehensive multiscale understanding of structure-property relationships in these complex materials (George & Brow, 2015; Lu, et al 2021).

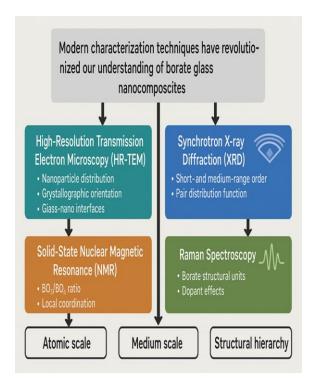


Figure 3: Characterization techniques of borate glass nanocomposites.

Recent developments in correlative microscopy approaches, where multiple techniques are applied to precisely the same sample region, are particularly powerful for bridging the gap between local atomic arrangements and macroscopic material behavior. These advanced characterization methods have become indispensable tools for both fundamental research and industrial development of borate glass nanocomposites, enabling precise quality control and accelerating the discovery of new material compositions with optimized performance characteristics.

2.3. Computational Modeling Approaches

Computational modeling approaches have emerged as indispensable tools for understanding and predicting the complex behavior of borate glass nanocomposites, complementing experimental characterization techniques. Molecular dynamics (MD) simulations, employing either classical force fields or reactive potentials, provide atomic-scale insights into nanoparticle-glass interactions by tracking the temporal evolution of interface formation, structural relaxation processes, and stress distribution patterns during thermal cycling. These simulations reveal how nanoparticle surface chemistry influences the local glass network structure, particularly the BO₃/BO₄ distribution near interfaces, and can predict optimal processing conditions to prevent nanoparticle agglomeration (Liu et al, 2021; Ohkubo et al, 2024). Density functional theory (DFT) calculations enable precise determination of dopant incorporation energies. electronic structure modifications, and thermodynamic stability of various coordination environments, offering fundamental explanations for experimentally observed phenomena such as the boron anomaly or dopant clustering tendencies (Rada et al, 2008). Advanced DFT approaches, including hybrid functionals and DFT+U methods, are particularly valuable for modeling transition metal and rare earth dopants, accurately capturing their electronic transitions and magnetic properties (Subedi & Tutchton, 2024). Monte Carlo (MC) modeling, especially when combined with machine learning potentials, efficiently simulates nanoparticle dispersion statistics across large length scales, predicting percolation thresholds and spatial distribution patterns that govern properties like electrical conductivity and optical scattering. Recent progress in multiscale modeling frameworks that seamlessly integrate these approaches—combining quantum mechanical accuracy at interfaces with mesoscale particle distribution statistics—has enabled the virtual design of nanocomposites with tailored properties before experimental synthesis (Swenson et al, 1998). These computational tools are increasingly being deployed in high-throughput screening of composition spaces and as digital twins for real manufacturing processes, significantly accelerating the development cycle for advanced borate glass nanocomposites.

2.4. Structure-Property Relationships

The engineered nanostructure of borate glass nanocomposites establishes fundamental structureproperty relationships that govern their performance in advanced applications. Thermal properties exhibit marked improvements, with nanoparticle loading typically increasing the glass transition temperature (T_g) depending on filler content and interface quality, as the constrained mobility of polymer-like borate chains near nanoparticle surfaces creates additional energy barriers for structural relaxation (Thakur et al. 2015). Mechanical behavior demonstrates particularly dramatic enhancements, where welldispersed nanoparticles can simultaneously increase hardness and fracture toughness through multiple reinforcing mechanisms: nanoscale fillers act as obstacles for crack propagation, forcing cracks to bow around particles or follow energetically unfavorable paths, while the residual stress fields surrounding nanoparticles promote crack deflection

(Januchta et al, 2019). Optical and branching characteristics become highly tunable through quantum confinement effects in semiconductor nanoparticles and plasmonic interactions in metallic inclusions, enabling precise control over bandgap and refractive index while maintaining excellent transparency in selected wavelength windows (Al-Hadeethi, 2022). Chemical durability improvements stem from nanoparticle-induced modifications to surface chemistry, where fillers either consume reactive non-bridging oxygens or create diffusion barriers that reduce leaching rates by orders of magnitude in corrosive environments (Abodunrin et al, 2023). These property enhancements frequently exhibit nonlinear relationships with nanoparticle content, typically following percolation-like behavior where dramatic improvements occur once a critical interfacial network forms throughout the glass matrix. Recent studies reveal that the most significant property modifications occur not from the nanoparticles themselves, but from the modified glass structure in the 2-5 nm interfacial zones surrounding each particle, where the borate network adapts to minimize interfacial energy through coordination changes and density variations. This understanding has led to the development of "interface-engineered" nanocomposites where surface-functionalized nanoparticles are used to deliberately create specific interfacial structures that optimize multiple properties simultaneously, representing a paradigm shift from conventional composition-based design to interfacecontrolled material engineering.

3. Preparation Techniques

Several techniques exist for fabricating glass nanocomposites. Melt-quenching is widely used for its simplicity, followed by controlled heat treatment to induce crystallization. Other methods include Sol-Gel, Chemical Vapor Deposition (CVD), Physical Vapor Deposition (PVD), and Hot Pressing. Each technique offers distinct advantages in tailoring the microstructure and performance of the resulting materials.

3.1 Melt Quenching

It is a simple way for glass nanocomposites preparation, and the most popular for its simplicity. The chemical ingredients weighted and mixed in an Aluminum crucible that withstand high temperatures. Components may be heated at low temperature before melting for good mixing and to ensure that some components do not evaporate during melting. Then the mixture melts by using electric furnace at high temperature for a specific period of time. During

melting the glass in liquid state so the melt poured into the mold that shapes the samples in the required shape for different measurements (See Fig. 5). Annealing at temperature near the glass transition temperature may be needed for remove residual thermal stress (Kashif et al, 2012; Taha et al, 2024). Sometimes heat treatment is required for growing nanocrystals in glass matrix, where the samples reheated for a specific time at temperature above Tg and blow Tm. Samples may be reheated at low temperature below Tg for a period of time to remove the undesirable mechanical stresses, and to ensure that the physical properties are uniform throughout the glass (El-Rabaie et al, 2014a; 2014b). Fig. 4 shows ischemic diagram of melt quenching technique.

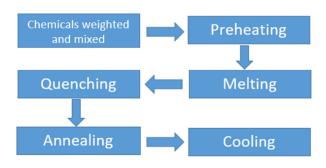


Figure 4: Schematic diagram of melt-quenching technique.

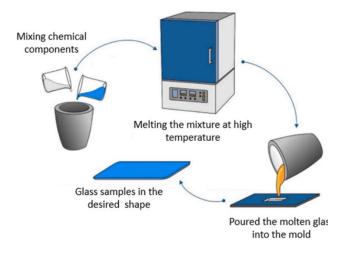


Figure 5: Melt-quenching technique.

4.2 Sol-Gel Method

Sol-Gel process based on the hydrolysis and polymerization of metal alkoxide precursors of alumina, silica, titania, zirconia, as well as other oxides to form a solution. The solution stirred well by using magnetic stirrer to form high viscosity solution, the solutions of precursors are reacted to form irreversible gels (ageing), that dry for water and other solvents removing by continuous heating at specific temperature for a certain. The last step is annealing at high temperature (calcining) to form a dense glass (see Fig. 6). This method can be carried out at low temperatures as compared melt-quenching technique, which makes it compatible with organic polymers. This method results in high purity materials, and allows fine control over the composition and microstructure. Can be applied for optical devises, electronics, and catalysis (Mukherjee, 1980; Zheng & Boccaccini, 2017; Roman et al, 2003).

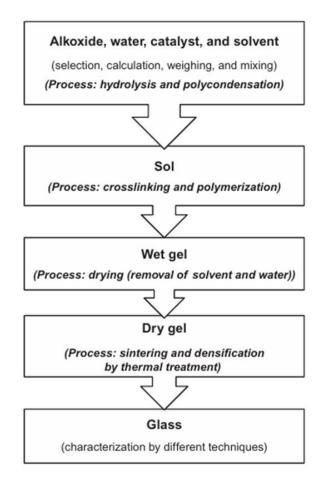


Figure 6: Schematic diagram of glass preparation by sol-gel technique.

4.3 Chemical Vaporization Deposition (CVD)

This method uses chemical reaction for solid material deposition from a vapor on heated substrate surface. The procedures are shown in the figure 7. High purity, uniform properties, thin film with controlled

thickness can be produced because CVD has excellent throwing power. CVD can localize or select the deposition pattern. CVD with cold-wall reactors reduce contaminations on the wall than hot-wall reactors. CVD use for thin film production in different fields (microelectronics, optical fibers, solar energy conversion, and semiconductor lasers), for amorphous, glassy thin film fabrication (Gordon, 1997; Karches et al, 2002). CVD has three main components as shown in the figure 8.

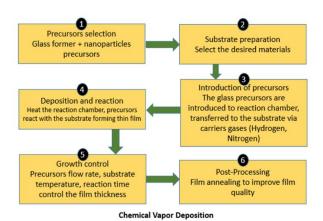


Figure 7: Schematic diagram of chemical vapor deposition procedures.

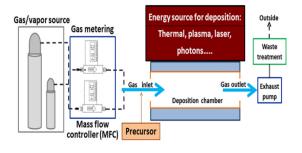


Figure 8: Chemical vapor deposition main components.

3.4 Physical Vaporization Deposition (PVD)

Thin film glass nanocomposites formation by physical transfer of materials from the source to the substrate, as shown in the diagram. Crystalline, or amorphous structure can be produced, this method proceeds under vacuum or low gas pressure, where the film formation based on pure physical condensation, or chemical reactions. This method is the gold standard for the preparation of multilayered structures, graded films, free-standing structures, etc. PVD is used for the improvement of wear resistance, oxidation resistance, lubricity, surface roughness,

cutting tools, and sliding characteristics, so it used for coating tools (See Fig. 9). The nano-layer thickness and composition can be precisely controlled by this method. Different processes can by used such as thermal, electron beam, pulsed laser, direct current, and radiofrequency magnetron sputtering in this method (Mohri et al, 2017; Cué-Sampedro et al, 2019).

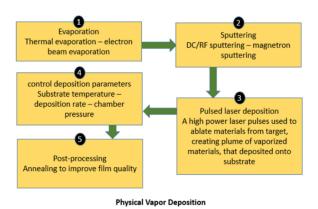


Figure 9: Schematic diagram of physical vapor deposition procedures.

3.5 Hot Pressing

Hot pressing method used for glass nanocomposites fabrication, based on the application of high temperature and high pressure to form high dense materials. This method produces high mechanical properties material, controls microstructure, and has uniform disruption of nanoparticles. This method can be used for densification glass ceramic products. Hot pressing method used for optoelectronics, energy storage, and biomedical devices (Decottignies et al, 1978; Matamoros-Veloza et al 2008). Figure 10 shows the details of glass ceramic preparation via hot pressing method.

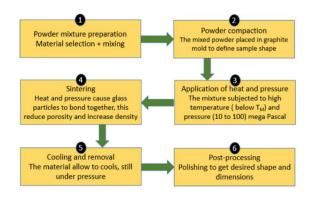


Figure 10: Schematic diagram of Hot-pressing procedures.

4. Dielectric Properties of Borate Glasses

Borate glass nanocomposites exhibit highly tunable dielectric properties, making them promising candidates for advanced electronic applications; including high-energy-density capacitors, microwave substrates, and solid-state electrolytes. The dielectric behavior of these materials is governed by complex interactions between the borate glass matrix and embedded nanoparticles, leading to several key phenomena (Heikal et al, 2025; Biswas et al, 2024).

4.1 Interfacial Polarization and Enhanced Permittivity

The incorporation of metal oxides (e.g., TiO₂, BaTiO₃, or conductive fillers) introduces interfacial polarization (Maxwell-Wagner-Sillars effect), where charge carriers accumulate at the glass-nanoparticle boundaries under an applied electric field. This effect significantly increases the effective dielectric constant (ε'), with some nanocomposites achieving ε' values exceeding 100 at low frequencies—orders of magnitude higher than undoped borate glasses. The magnitude of this enhancement depends on nanoparticle conductivity, volume fraction, and dispersion quality, with percolation effects playing a critical role near conductive filler thresholds (Paramesh & Varma, 2013; Li et al, 2018; Salem et al, 2011).

4.2 Low Dielectric Loss for High-Frequency Applications

Optimized nanocomposites demonstrate remarkably low dielectric loss (tan $\delta < 0.01$ at MHz-GHz frequencies), a crucial requirement for 5G/6G communication devices and high-speed electronics. This is achieved through precise control of nanoparticle composition (e.g., using high-purity oxides) and glass network stabilization (e.g., reducing OH- content). Recent studies show that core-shell nanoparticle architectures can further suppress loss mechanisms by preventing direct current leakage paths while maintaining high permittivity (Tripathi et al, 2019).

4.3 Ionic Conductivity in Solid-State Battery Applications

Lithium-doped borate glass nanocomposites (e.g., Li₂O-B₂O₃ with LLZO or LATP nanoparticles) exhibit exceptional ionic conductivity (>10⁻³ S/cm at room temperature) due to synergistic effects: the borate glass provides a percolating Li⁺ conduction pathway, while embedded nanoparticles create space-charge regions that enhance ion mobility along interfaces. Advanced nanocomposite designs now incorporate 3D-connected ceramic nano frameworks to achieve both high conductivity and mechanical

stability for next-generation solid-state batteries (Graça et al, 2008; Rai & Kundu, 2023; Dongare 2017).

4.4 Frequency-Dependent Behavior and Temperature Stability

The dielectric response of these materials shows strong frequency dispersion, transitioning from interfacial polarization dominance at low frequencies to dipolar/ionic relaxation at higher frequencies. Carefully engineered compositions maintain stable dielectric properties across broad temperature ranges (-50 °C to 300 °C), enabling applications in harsh environments. Emerging research focuses on nanocomposites with negative permittivity for metamaterials and on-demand dielectric switching behavior for adaptive electronics (Tretyakov, 2001; Tallman, 2020).

5. Radiation Shielding Applications

Borate-based glass-ceramics are emerging as alternatives to lead and concrete for gamma-ray shielding due to their non-toxic, transparent, and thermally stable nature. The shielding efficiency is evaluated using parameters like linear attenuation coefficient, HVL, Zeff, and RPE. Adding heavy metal oxides enhances these properties, making them suitable for medical and nuclear safety applications. Today ionization radiation is used for many applications such as radiation thereby, medical imaging, agriculture, and sterilization. And the exposure to ionization radiation causes health risks. The need to reduce exposure to radiation becomes a very important role occupational health safety and radiation protection (Al-Buriahi, 2023; Al-Hadeethi, 2022; Al-Buriahi, 2025, Bagheri & Shirmardi, 2021).

One of the most important issues in radiation safety is radiation shielding, where the distance, time, and shielding are the basic of ALARA principle for radiation protection. In medicine, radiation thereby, sterilization, and space application radiation shielding material is needed. The most essential requirements for radiation shielding material are high atomic number, high density, and proper thickness to be easy reconstructed and low cost in addition to must be environmentally safe (Mansouri et al, 2020). The most common radiation shielding materials are lead and concrete, but lead has low mechanical properties in addition to toxicity, and concrete is heavy, opaque, and losses water content when exposed to radiation. So, using glasses for radiation shielding material become widespread due to its important feature such as transparency, chemical stability, thermal stability, easy manufacture and features manipulating, low

cost, low weight, and biocompatible (Al-Buriahi, 2025).

Glass in contrast other radiation shielding materials characterized by transparency and water resistance than heavy, opaque, and hazard materials (lead, and concrete) (Saudi et al, 2020). Concrete is used as a radiation shielding, but concrete losses its water content when exposed to nuclear radiation. Some applications need the transparency of the glasses such as nuclear medicine laps, X-ray and computed tomography scanner's windows, moreover its ability of glasses to absorb neutrons and gamma rays, so glasses regarded as new generation of radiation shielding applications (Abdullah et al, 2022; Azreen et al, 2018).

Borate-based glass nanocomposites have emerged as highly promising materials for advanced radiation shielding applications due to their structural flexibility, compositional tunability, and ability to incorporate high-Z dopants and nanofillers. The integration of heavy metal oxides such as Bi₂O₃ (Biradar et al, 2024; Mhareb, 2023) and Gd₂O₃ (Kilic, 2023) significantly enhances gamma and neutron attenuation by increasing the mass attenuation coefficient while maintaining advantages like low toxicity, transparency, and lightweight nature compared to conventional lead-based shields. Nanoparticles further improve shielding through interface-driven effects, including enhanced scattering, bond rearrangements, and densification at the glass-nanoparticle boundaries. Monte Carlo simulations have been widely employed to validate and optimize the shielding performance of these composites. Future directions emphasize the development of multifunctional materials that combine radiation shielding with smart features such as real-time monitoring via scintillation from rareearth dopants (Eu³⁺ (Kilic et al, 2021; Katubi et al, 2024), Tb³⁺), and adaptive shielding using photochromic or piezoelectric components for ondemand protection and sensing. Additionally, the review highlights the importance of sustainable design, promoting eco-friendly, recyclable borate glass systems, which offer excellent shielding efficiency without relying on hazardous substances. Together, these advances establish borate glass nanocomposites as next-generation candidates for protective technologies in medical, nuclear, and aerospace environments.

6. Challenges and Future Perspectives

Despite the remarkable progress in borate-based glass nanocomposites, several critical challenges must be addressed to facilitate their widespread industrial adoption. One of the most persistent issues is

achieving uniform nanoparticle dispersion without agglomeration, which currently requires precise control over synthesis parameters (temperature, viscosity, cooling rates) and often leads to batch-tobatch inconsistencies. Additionally, high production costs associated with rare-earth dopants (e.g., Gd₂O₃) and specialized nano-processing techniques (e.g., spark plasma sintering) limit large-scale commercialization. Scaling up laboratory synthesis methods while maintaining nanostructural precision remains a significant hurdle, particularly for glasses requiring controlled crystallization or hybrid organicinorganic phases.

Future research on borate glass nanocomposites should prioritize the development of multifunctional materials that synergistically combine radiation shielding with smart responsive properties, opening new possibilities for advanced applications. A key direction involves integrating luminescent rare-earth dopants (Eu³⁺, Tb³⁺, Dy³⁺) to create self-diagnostic shields capable of real-time radiation monitoring through scintillation effects, where the intensity and wavelength of emitted light correlate directly with radiation dose and type. Simultaneously, incorporating photochromic (e.g., AgCl-doped) or electrochromic (e.g., WO₃-based) components could enable adaptive shielding systems that dynamically modulate their opacity and attenuation properties in response to changing radiation fields or electrical stimuli, particularly valuable for spacecraft and variable-intensity medical imaging environments. Furthermore, embedding piezoelectric nanoparticles (e.g., ZnO, BaTiO₃) within the glass matrix presents an innovative approach to self-powered radiation detection, where the energy from incident radiation generates measurable electrical signals through the piezoelectric effect, eliminating the need for external power sources in detector applications. These multifunctional designs require precise nanoscale engineering to prevent property interference - for instance, ensuring scintillation centers don't quench piezoelectric responses - potentially achieved through core-shell nanoparticle architectures or spatially graded compositions. Such advanced materials could revolutionize radiation protection by merging shielding, sensing, and adaptive response into single, compact systems for nuclear facilities, space habitats, and portable medical devices.

The development of eco-friendly and sustainable borate glass nanocomposites represents a critical research direction, focusing on replacing conventional toxic heavy metals (Pb, Cd) with high-performance, environmentally benign alternatives while maintaining superior shielding capabilities. Lead-free gamma shielding compositions based on

Bi₂O₃-WO₃-TeO₂ systems have shown particular promise, as bismuth's high atomic number (Z=83) provides excellent photon attenuation comparable to lead, while tungsten oxide (WO₃) enhances mechanical stability and tellurium oxide (TeO₂) improves glass-forming ability, creating a balanced ternary system suitable for medical and nuclear applications. For neutron absorption, innovative boron-rich polymers and bio-glasses derived from sustainable precursors (e.g., borosilicate waste or biologically compatible compounds) lightweight, non-toxic alternatives to traditional cadmium-based shields, with the added advantage of tunable hydrogen content for enhanced moderation capabilities. A crucial parallel effort involves designing fully recyclable nanocomposites through strategic selection of components with similar melting points and chemical compatibility, enabling closed-loop material reprocessing without property degradation; this includes developing nanoparticle coatings that prevent agglomeration during remelting and optimizing glass matrices for easy separation of constituent materials. These sustainable approaches not only address environmental and health concerns but also open new opportunities for cost-effective production and end-of-life material management, making radiation shielding technologies more accessible for large-scale civil applications while meeting increasingly stringent global regulations on hazardous substances.

Recent advances in manufacturing techniques are revolutionizing the production of borate glass nanocomposites, enabling unprecedented control over material properties and structure. Sol-gel processing has emerged as a powerful low-temperature (300-600 °C) alternative to conventional melt-quenching. allowing superior nanoparticle homogeneity through molecular-level mixing of precursors minimizing agglomeration risks. For consolidation, field-assisted sintering techniques (FAST/SPS) combine high heating rates with uniaxial pressure to achieve >99% theoretical density at reduced temperatures, preserving nanoscale features that would degrade in conventional furnaces and enabling unique glass-ceramic nanocomposites with tailored crystallinity. Additive manufacturing (3D printing) of borate glasses, particularly via direct ink writing or selective laser melting, permits the fabrication of functionally graded shields with spatially varying compositions optimized for multi-spectral radiation attenuation - an impossibility with traditional forming methods. Most transformatively, AI-driven materials design is accelerating development cycles, where machine learning algorithms trained on vast datasets of composition-structure-property relationships can predict optimal dopant combinations and processing

parameters, potentially identifying nanocomposite formulations with exceptional shielding efficiency, mechanical robustness, and optical clarity. These advanced manufacturing approaches collectively address longstanding challenges in scalability, reproducibility, and performance optimization, transitioning borate glass nanocomposites from laboratory curiosities to industrially viable next-generation materials.

Future advancements in borate glass nanocomposites must prioritize enhanced performance metrics to meet the rigorous demands of cutting-edge applications. A critical focus lies in improving toughness through mechanical nanoscale reinforcement strategies, such as incorporating carbon nanotubes or graphene oxide platelets to crack-bridging networks that resist embrittlement under prolonged radiation exposure, while maintaining the composite's attenuation capabilities. Simultaneously, extending thermal stability to 1000 °C requires innovative approaches like refractory nanoparticle additions (ZrO₂, HfO₃) and controlled crystallization to develop stable glassceramic nanocomposites capable of withstanding extreme nuclear reactor environments without devitrification or property degradation. For medical imaging applications, optimizing optical clarity demands precise engineering of nanoparticle size (<20 nm) and interfacial chemistry to minimize light scattering while preserving shielding efficacy, potentially through advanced polishing techniques resonant plasmonic nanostructures selectively block ionizing radiation while transmitting visible wavelengths. These performance enhancements must be achieved synergistically, requiring fundamental advances in understanding radiation-induced structural changes, temperature phase stability, and light-matter interactions at the nanoscale - challenges that will drive the next generation of multifunctional shielding materials for aerospace, energy, and healthcare technologies.

The future of borate glass nanocomposites lies in smart, adaptive materials that respond to radiation flux, self-monitor damage, and even self-heal microcracks. Advances in nanoscale interface engineering and multi-material integration (e.g., glass-ceramic hybrids) could unlock unprecedented performance, making these composites viable for space exploration, fusion reactors, and next-gen medical therapy systems. Collaborative efforts between material scientists, engineers, and industry stakeholders will be crucial to transition these materials from lab-scale breakthroughs to real-world solutions.

7. Conclusions

Borate-based glass nanocomposites represent a versatile class of materials with significant potential in dielectric and radiation shielding applications. Recent advancements in nanostructuring and doping strategies have enhanced their structural, electrical, and protective properties. Further research is needed to optimize these materials for commercial use, particularly in medical imaging, nuclear safety, and optoelectronic devices.

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References

- Abdullah, M. A. H., Rashid, R. S. M., Amran, M., Hejazii, F., Azreen, N. M., Fediuk, R., ... & Idris, M. I. (2022). Recent trends in advanced radiation shielding concrete for construction of facilities: materials and properties. Polymers, 14(14), 2830.
- Abodunrin, O. D., El Mabrouk, K., & Bricha, M. (2023). A review on borate bioactive glasses (BBG): effect of doping elements, degradation, and applications. Journal of Materials Chemistry B, 11(5), 955-973.
- AbuAlRoos, N. J., Amin, N. A. B., & Zainon, R. (2019). Conventional and new lead-free radiation shielding materials for radiation protection in nuclear medicine: A review. Radiation Physics and Chemistry, 165, 108439.
- Adamiv, V. T., Bolesta, I. M., Burak, Y. V.,
 Gamernyk, R. V., Karbovnyk, I. D., Kolych, I.
 I., ... & Teslyuk, I. M. (2014). Nonlinear optical properties of silver nanoparticles prepared in Ag doped borate glasses. Physica B: Condensed Matter, 449, 31-35.
- Ahmmad, S. K., Alsaif, N. A., Shams, M. S., El-Refaey, A. M., Elsad, R. A., Rammah, Y. S., & Sadeq, M. S. (2022). Machine learning density

- prediction and optical properties of calcium boro-zinc glasses. Optical Materials, 134, 113145.
- Al-Buriahi, M. S. (2023). Radiation shielding performance of a borate-based glass system doped with bismuth oxide. Radiation Physics and Chemistry, 207, 110875.
- Al-Buriahi, M. S., Alsaiari, N. S., Baskin, M. U., & Olarinoye, I. O. (2025). Recent progress in the radiation shielding performance of common glass systems: The roles of different class of modifiers. Journal of Radiation Research and Applied Sciences, 18(1), 101264.
- Al-Buriahi, M. S., Sriwunkum, C., Arslan, H., Tonguc, B. T., & Bourham, M. A. (2020). Investigation of barium borate glasses for radiation shielding applications. Applied Physics A, 126(1), 68.
- Alderman, O. L. G. (2018). Borate melt structure: A short review. Physics and Chemistry of Glasses-European Journal of Glass Science and Technology Part B, 59(1), 1-10.
- Al-Hadeethi, Y., Sayyed, M. I., Barasheed, A. Z., Ahmed, M., & Elsafi, M. (2022). Fabrication of lead free borate glasses modified by bismuth oxide for gamma ray protection applications. Materials, 15(3), 789.
- Al-Hadeethi, Y., Sayyed, M. I., Barasheed, A. Z., Ahmed, M., & Jagannath, G. (2022). Nanosecond nonlinear optical, optical limiting and structural properties of Eu³⁺ activated antimony sodium borate glasses embedded with silver nanoparticles: Effect of heat treatment. Optical Materials, 125, 112106.
- Alhakami, M. H., Abouhaswa, A. S., Althubiti, N. A., & Taha, T. A. M. (2025). Investigation of optical and radiation shielding properties in bismuth oxide-doped barium borate glasses. Nuclear Engineering and Technology, 57(9), 103633.
- Ande, A., Bhemarajam, J., Kanneboina, V., & Prasad, M. (2021). Influence of Ag nano particles on spectroscopic and luminescence properties of Dy³⁺ doped borate glasses. Journal of Non-Crystalline Solids, 559, 120702.
- Azreen, N. M., Rashid, R. S., Haniza, M., Voo, Y. L., & Amran, Y. M. (2018). Radiation shielding of ultra-high-performance concrete with silica

- sand, amang and lead glass. Construction and Building Materials, 172, 370-377.
- Bagheri, R., & Shirmardi, S. P. (2021). Gamma-ray shielding studies on borate glasses containing BaO, Bi₂O₃, and PbO in different concentrations. Radiation Physics and Chemistry, 184, 109434.
- Biradar, S., Chandrashekara, M. N., Dinkar, A., Devidas, G. B., Bennal, A. S., Rajaramakrishna, R., & Sayyed, M. I. (2024). Synergistic optimization of physical, thermal, structural, mechanical, optical and radiation shielding characteristics in borate glasses doped with Bi₂O₃. Optical Materials, 155, 115815.
- Biswas, D., Mandal, A., Mondal, R., Kabi, S., & Modak, N. (2024). Influence of lithium-ion doping on electrical and dielectric properties of zinc-borate-phosphate glass for improved energy storage. Journal of Alloys and Compounds, 1005, 176170.
- Button, D. P., Tandon, R., King, C., Velez, M. H., Tuller, H. L., & Uhlmann, D. R. (1982). Insights into the structure of alkali borate glasses. Journal of Non-Crystalline Solids, 49(1-3), 129-142.
- Chakrabarti, A., Menon, S., Tarafder, A., & Molla, A. R. (2022). Glass-ceramics: A Potential Material for Energy Storage and Photonic Applications. In Glasses and Glass-Ceramics: Advanced Processing and Applications (pp. 265-304). Singapore: Springer Nature Singapore.
- Colak, S. C. (2017). Role of titanium ions on the optical and thermal properties of zinc borate glass doped with TiO₂. Physics and Chemistry of Glasses-European Journal of Glass Science and Technology Part B, 58(2), 41-48.
- Cué-Sampedro, R., Hernández-Hernández, R., & Garza, L. L. (2019). Research on UV attenuating nanocoatings on glass using PVD by sputtering. Materials Today: Proceedings, 13, 349-352.
- Decottignies, M., Phalippou, J., & Zarzycki, J. (1978). Synthesis of glasses by hot-pressing of gels. Journal of Materials Science, 13(12), 2605-2618.
- Deepika, Dixit, M., Singh, H., Attia, M. S., & Amin, M. A. (2021). Recent innovations in properties

- of nanostructured glasses and composites. Journal of Experimental Nanoscience, 16(1), 180-211.
- Dongare, D. T. (2017). Enhancement in Electrical Conductivity of Lithium Alumina Borate Glasses. International Journal of Pure And Applied Research, 2(2), 93-96.
- Eckert, H. (2018). Structural characterization of bioactive glasses by solid state NMR. Journal of Sol-Gel Science and Technology, 88(2), 263-295.
- Edwards, T., Endo, T., Walton, J. H., & Sen, S. (2014). Observation of the transition state for pressure-induced BO₃→ BO₄ conversion in glass. Science, 345(6200), 1027-1029.
- El-Rabaie, S., Taha, T. A., & Higazy, A. A. (2014). PbTe quantum dots formation in a novel germanate glass. Journal of alloys and compounds, 594, 102-106.
- El-Rabaie, S., Taha, T. A., & Higazy, A. A. (2014). Synthesis and characterization of CdS nanocrystals embedded in germanate glasses. Applied Nanoscience, 4(2), 219-226.
- Eskalen, H., Kavgaci, M., Kavun, Y., Yaykaşli, H., Gülpak, H., & Gök, M. (2025). CuO nanoparticle doping in barium borate glasses: effects on density prediction via machine learning, optical, mechanical, thermal, and radiation protection properties. Journal of Materials Science: Materials in Electronics, 36(18), 1-16.
- Eskalen, H., Kavun, Y., & Kavgacı, M. (2023).

 Preparation and study of radiation shielding features of ZnO nanoparticle reinforced borate glasses. Applied Radiation and Isotopes, 198, 110858.
- Fan, L., Wei, S., Li, S., Li, Q., & Lu, Y. (2018). Recent progress of the solid-state electrolytes for high-energy metal-based batteries. Advanced Energy Materials, 8(11), 1702657.
- Fan, S., Wu, G., Zhang, Y., Chai, G., Ma, Z., Qiu, J., & Dong, G. (2014). Novel visible emission and mechanism investigation from PbS nanoclusters-doped borosilicate glasses. Journal of the American Ceramic Society, 97(1), 173-178.
- Fayaz, M., Ali, S., Bibi, S., Iqbal, Y., Ali, M., Zaman, F., ... & Kaewkhao, J. (2023). Luminescence and energy transfer mechanism in Ce³⁺ and

- Gd³⁺ ions in bismuth borate glass. Ceramics International, 49(15), 24690-24695.
- Fonseca, J. (2022). Nanoparticles embedded into glass matrices: glass nanocomposites. Frontiers of Materials Science, 16(3), 220607.
- Funabiki, F., Matsuishi, S., & Hosono, H. (2011). Ligand field modification around Cu²⁺ ions in sodium borate glass by codoping. The Journal of Physical Chemistry A, 115(20), 5081-5088.
- George, J. L., & Brow, R. K. (2015). In-situ characterization of borate glass dissolution kinetics by μ-Raman spectroscopy. Journal of Non-Crystalline Solids, 426, 116-124.
- Gordon, R. (1997). Chemical vapor deposition of coatings on glass. Journal of Non-Crystalline Solids, 218, 81-91.
- Graça, M. P. F., da Silva, M. F., Sombra, A. S. B., & Valente, M. A. (2008). The dielectric behavior of a thermoelectric treated B₂O₃–Li₂O–Nb₂O₅ glass. Journal of non-crystalline solids, 354(29), 3408-3413.
- Gurushantha, K., Swetha, B. N., Kolavekar, S. B., Pramod, A. G., Almuqrin, A. H., Sayyed, M. I., ... & Keshavamurthy, K. (2023). Nonlinear optical properties of gold nanoparticles containing borate glasses at spectral region of surface plasmon resonance. Materials Today Communications, 35, 106032.
- Hannon, A. C. (2015). Neutron diffraction techniques for structural studies of glasses. Modern glass characterization, 1-83.
- Hayes, T. M., Persans, P. D., Filin, A., & Peng, C. (2004). Bonding changes during the growth of CdSe nanoparticles in glass. Journal of noncrystalline solids, 349, 35-37.
- Heikal, S., El-Bassyouni, G. T., & Hamzawy, E. M. (2025). Characterization, Dielectric and AC Electrical Properties of B₂O₃-Al₂O₃-BaO-x TiO₂ (x= 0, 10, 15, and 20) Thermally Treated Glass. Journal of Non-Crystalline Solids, 666, 123687.
- Intachai, N., Wantana, N., Kaewjaeng, S., Chaiphaksa, W., Cheewasukhanont, W., Htun, K. T., ... & Kaewkhao, J. J. R. P. (2022). Effect of Gd₂O₃ on radiation shielding, physical and optical properties of sodium borosilicate glass system. Radiation Physics and Chemistry, 199, 110361.

- Ivanova, O. S., Ivantsov, R. D., Edelman, I. S., Petrakovskaja, E. A., Velikanov, D. A., Zubavichus, Y. V., ... & Stepanov, S. A. (2016). Identification of ε-Fe₂O₃ nano-phase in borate glasses doped with Fe and Gd. Journal of Magnetism and Magnetic Materials, 401, 880-889.
- Jagannath, G., Eraiah, B., NagaKrishnakanth, K., & Rao, S. V. (2018). Linear and nonlinear optical properties of gold nanoparticles doped borate glasses. Journal of Non-Crystalline Solids, 482, 160-169.
- Januchta, K., Youngman, R. E., Jensen, L. R., & Smedskjaer, M. M. (2019). Mechanical property optimization of a zinc borate glass by lanthanum doping. Journal of Non-Crystalline Solids, 520, 119461.
- Kamitsos, E. I. (2025). Borate glass structure by vibrational spectroscopy: a review on borate speciation and metal ion–site interactions. Physics and Chemistry of Glasses-European Journal of Glass Science and Technology Part B, 66(1), 1-38.
- Kamitsos, E. I., Varsamis, C. P. E., & Vegiri, A. (2020). Spectroscopic studies of mobile cations in glass. Physics and Chemistry of Glasses-European Journal of Glass Science and Technology Part B, 61(3), 107-122.
- Karches, M., Morstein, M., Von Rohr, P. R., Pozzo, R. L., Giombi, J. L., & Baltanás, M. A. (2002). Plasma-CVD-coated glass beads as photocatalyst for water decontamination. Catalysis Today, 72(3-4), 267-279.
- Kashif, I., Soliman, A. A., Sakr, E. M., & Ratep, A. (2012). Effect of different conventional melt quenching technique on purity of lithium niobate (LiNbO₃) nano crystal phase formed in lithium borate glass. Results in Physics, 2, 207-211.
- Kato, T., Lodesani, F., & Urata, S. (2024). Boron coordination and three-membered ring formation in sodium borate glasses: A machine-learning molecular dynamics study. Journal of the American Ceramic Society, 107(5), 2888-2900.
- Katubi, K. M., Alrowaili, Z. A., Olarinoye, I. O., & Al-Buriahi, M. S. (2024). Optical properties and radiation protection applications of B₂O₃:

- Na₂O: PbO: Tb₂O₃: Bi₂O₃ glass system. Optical and Quantum Electronics, 56(4), 625.
- Kilic, G., Issa, S. A., Ilik, E., Kilicoglu, O., & Tekin,
 H. O. (2021). A journey for exploration of Eu₂O₃ reinforcement effect on zinc-borate glasses: Synthesis, optical, physical and nuclear radiation shielding properties.
 Ceramics International, 47(2), 2572-2583.
- Kojima, S. (2020). Mixed-alkali effect in borate glasses: thermal, elastic, and vibrational properties. Solids, 1(1), 16-30.
- Lepry, W. C., & Nazhat, S. N. (2015). Highly bioactive sol-gel-derived borate glasses. Chemistry of Materials, 27(13), 4821-4831.
- Li, T., Segawa, H., & Ohashi, N. (2018). Sintering behavior and dielectric properties of BaTiO₃ added with BaO-Bi₂O₃-B₂O₃ glass phase. Ceramics International, 44(11), 13004-13010.
- Liu, H., Fu, Z., Yang, K., Xu, X., & Bauchy, M. (2021). Machine learning for glass science and engineering: A review. Journal of Non-Crystalline Solids, 557, 119419.
- Liu, S., Shen, B., Hao, H., & Zhai, J. (2019). Glass– ceramic dielectric materials with high energy density and ultra-fast discharge speed for high power energy storage applications. Journal of Materials Chemistry C, 7(48), 15118-15135.
- Lower, N. P., McRae, J. L., Feller, H. A., Betzen, A. R., Kapoor, S., Affatigato, M., & Feller, S. A. (2001). Physical properties of alkaline-earth and alkali borate glasses prepared over an extended range of compositions. Journal of non-crystalline solids, 293, 669-675.
- Lu, X., Deng, L., Du, J., & Vienna, J. D. (2021). Predicting boron coordination in multicomponent borate and borosilicate glasses using analytical models and machine learning. Journal of Non-Crystalline Solids, 553, 120490.
- Mansouri, E., Mesbahi, A., Malekzadeh, R., & Mansouri, A. (2020). Shielding characteristics of nanocomposites for protection against X-and gamma rays in medical applications: effect of particle size, photon energy and nanoparticle concentration. Radiation and Environmental Biophysics, 59(4), 583-600.
- Marimuthu, K., Surendra Babu, S., Muralidharan, G., Arumugam, S., & Jayasankar, C. K. (2009). Structural and optical studies of Eu3+ ions in

- alkali borate glasses. Physica status solidi (a), 206(1), 131-139.
- Massot, M., Souto, S., & Balkanski, M. (1995). Short and medium range order in ternary borate glasses. Journal of non-crystalline solids, 182(1-2), 49-58.
- Matamoros-Veloza, Z., Rendón-Angeles, J. C., Yanagisawa, K., Cisneros-Guerrero, M. A., Cisneros-Guerrero, M. M., & Aguirre, L. (2008). Preparation of foamed glasses from CRT TV glass by means of hydrothermal hotpressing technique. Journal of the European Ceramic Society, 28(4), 739-745.
- Mhareb, M. H. A. (2023). Optical, Structural, Radiation shielding, and Mechanical properties for borosilicate glass and glass ceramics doped with Gd_2O_3 . Ceramics International, 49(22), 36950-36961.
- Mohri, M., Wang, D., Ivanisenko, J., Gleiter, H., & Hahn, H. (2017). Investigation of the deposition conditions on the microstructure of TiZrCuPd nano-glass thin films. Materials Characterization, 131, 140-147.
- Möncke, D., Tricot, G., Winterstein-Beckmann, A., Wondraczek, L., & Kamitsos, E. I. (2015). On the connectivity of borate tetrahedra in borate and borosilicate glasses. Physics and Chemistry of Glasses-European Journal of Glass Science and Technology Part B, 56(5), 203-211.
- Moriceau, J., Houizot, P., Pasturel, M., Guizouarn, T., & Rouxel, T. (2019). A magnetic glass matrix (ZnO-BaO-B₂O₃) particulate (Fe₃O₄) nanocomposite obtained by SPS. Journal of Non-Crystalline Solids, 514, 116-121.
- Mortazavi, S. M. J., Bevelacqua, J. J., Rafiepour, P., Sina, S., Moradgholi, J., Mortazavi, A., & Welsh, J. S. (2024). Lead-free, multilayered, and nanosized radiation shields in medical applications, industrial, and space research. In Advanced Radiation Shielding Materials (pp. 305-322). Elsevier.
- Mukherjee, S. P. (1980). Sol-gel processes in glass science and technology. Journal of Non-Crystalline Solids, 42(1-3), 477-488.
- Ohkubo, T., Sasaki, S., Masuno, A., & Tsuchida, E. (2024). Ab Initio Molecular Dynamics Study of Trivalent Rare Earth Rich Borate Glasses:

 Structural Insights and Formation

- Mechanisms. The Journal of Physical Chemistry B, 128(47), 11800-11813.
- Ohkubo, T., Tsuchida, E., Takahashi, T., & Iwadate, Y. (2016). Ab initio molecular dynamics simulations and GIPAW NMR calculations of a lithium borate glass melt. The Journal of Physical Chemistry B, 120(14), 3582-3590.
- Padmaja, G., & Kistaiah, P. (2009). Infrared and Raman spectroscopic studies on alkali borate glasses: evidence of mixed alkali effect. The Journal of Physical Chemistry A, 113(11), 2397-2404.
- Paramesh, G., & Varma, K. B. (2013). Structure-property correlation in BaO-TiO₂-B₂O₃ glasses: glass stability, optical, hydrophobic, and dielectric properties. International Journal of Applied Glass Science, 4(3), 248-255.
- Parthé, E. (2002). Calculation of the BO₃ triangle to BO₄ tetrahedron ratio in borates. Zeitschrift für Kristallographie-Crystalline Materials, 217(5), 179-200.
- Parthé, E. (2004). New examples of crystallized and amorphous borates where the ratio of BO₃ triangles to BO₄ tetrahedra can be calculated from the chemical formula. Journal of alloys and compounds, 367(1-2), 126-131.
- Pimentel, N. B., Mastelaro, V. R., M'peko, J. C., Martin, S. W., Rojas, S. S., & De Souza, J. E. (2018). Structural and electrical characterization of glasses in the Li₂O–CaO–B₂O₃ system. Journal of Non-Crystalline Solids, 499, 272-277.
- Rada, S., Culea, M., Neumann, M., & Culea, E. (2008). Structural role of europium ions in lead-borate glasses inferred from spectroscopic and DFT studies. Chemical Physics Letters, 460(1-3), 196-199.
- Rada, S., Pascuta, P., Rus, L., Rada, M., & Culea, E. (2012). Spectroscopic properties and ab initio calculations on the structure of erbium–zincborate glasses and glass ceramics. Journal of non-crystalline solids, 358(1), 30-35.
- Rai, K., & Kundu, S. (2023). B₂O₃ as an effective sintering aid in enhancing the lithium-ion conductivity of LiTa₂PO₈ solid electrolyte. Journal of Power Sources, 576, 233229.
- Ranasinghe, K. S., Singh, R., Day, D. E., Attenkofer, K., Stavitski, E., Quinn, L. A., ... & Duenas, A. (2019). Evidence of the coexistence of

- multivalence cerium oxide nano-particles in a sodium borate glass. Journal of Non-Crystalline Solids, 515, 75-81.
- Roman, J., Padilla, S., & Vallet-Regi, M. (2003). Sol— gel glasses as precursors of bioactive glass ceramics. Chemistry of materials, 15(3), 798-806.
- Salem, S. M., Antar, E. M., & Mohamed, E. A. (2011). Structural and dielectric properties of Li₂O–ZnO–BaO–B₂O₃–CuO glasses. Journal of Materials Science, 46(4), 1095-1102.
- Saudi, H. A., Hassaan, M. Y., Tarek, E., & Borham, E. (2020). Development of advanced, transparent radiation shielding glass possessing phosphate and lead ions in the glassy matrix. Journal of Optics, 49(4), 438-445.
- Seyfarth, B., Schade, L., Ullsperger, T., Matthäus, G., Tünnermann, A., & Nolte, S. (2018, February). Selective laser melting of borosilicate glass using ultrashort laser pulses. In Laser 3D Manufacturing V (Vol. 10523, pp. 52-57). SPIE.
- Shashikala, H. D., & Udayashankar, N. K. (2016). Investigation of mixed alkali effect on mechanical, structural and thermal properties of three-alkali borate glass system. Journal of Alloys and Compounds, 658, 996-1002.
- Shearer, A., Montazerian, M., Deng, B., Sly, J. J., & Mauro, J. C. (2024). Zirconia-containing glass-ceramics: From nucleating agent to primary crystalline phase. International Journal of Ceramic Engineering & Science, 6(2), e10200.
- Shilpa, S., Gurusiddappa, J., Kumar, G. R., Siddalingeshwara, B. P., & Hegde, G. S. (2025). A review on structural correlation to non-linear optical studies of REOs, HMOs and TMOs doped borate glasses. Journal of Optics, 1-27.
- Stone, C. E., Wright, A. C., Sinclair, R. N., Feller, S. A., Affatigato, M., Hogan, D. L., ... & Ehrt, D. (2000). Structure of bismuth borate glasses. Physics and chemistry of glasses, 41(6), 409-412.
- Subedi, K. N., & Tutchton, R. M. (2024). Atomic structure of uranium-incorporated sodium borosilicate glasses: An ab initio study. Physical Review B, 110(10), 104204.

- Svecova, B., Nekvindova, P., Stanek, S., Vytykacova, S., Mackova, A., Malinsky, P., ... & Spirkova, J. (2017). The effect of various silicate-glass matrixes on gold-nanoparticle formation. Ceramics–Silikáty, 61(1), 52-58.
- Swenson, J., Börjesson, L., & Howells, W. S. (1998). Structure of fast-ion-conducting lithium and sodium borate glasses by neutron diffraction and reverse Monte Carlo simulations. Physical Review B, 57(21), 13514.
- Taha, T. A. M., Alhakami, M. H., Althubiti, N. A., Saron, K. M. A., Dong, M., Ibrahim, M., & Saad, S. A. (2024). Melt-quenched zinc sodium borate glasses: Understanding the role of ZnO/Na₂O ratio in density, band gap, and radiation attenuation. Radiation Physics and Chemistry, 222, 111791.
- Tallman, T. N. (2020). The effect of thermal loading on negative permittivity in carbon nanofiber/silicone metacomposites. Materials Today Communications, 22, 100843.
- Thakur, V., Singh, A., Punia, R., Kaur, M., & Singh, L. (2015). Effect of BaTiO₃ on the structural and optical properties of lithium borate glasses. Ceramics International, 41(9), 10957-10965.
- Topper, B., & Möncke, D. (2022). Structure and properties of borate glasses. https://doi.org/10.1039/9781839164750-00162
- Tretyakov, S. A. (2001). Meta-materials with wideband negative permittivity and permeability. Microwave and Optical Technology Letters, 31(3), 163-165.
- Tripathi, P., Kumari, P., Mishra, V. K., Singh, R., Singh, S. P., & Kumar, D. (2019). Effect of PbO–B₂O₃–BaO–SiO₂ glass additive on dielectric properties of Ba_{0.5}Sr_{0.5}TiO₃ ceramics for radio-frequency applications. Journal of Physics and Chemistry of Solids, 127, 60-67.
- Urata, S., Tan, A. R., & Gómez-Bombarelli, R. (2024). Modifying ring structures in lithium borate glasses under compression: MD simulations using a machine-learning potential. Physical Review Materials, 8(3), 033602.
- Vijayakumar, R., Nagaraj, R., Suthanthirakumar, P., Karthikeyan, P., & Marimuthu, K. (2018). Silver (Ag) nanoparticles enhanced luminescence properties of Dy³⁺ ions in

- borotellurite glasses for white light applications. Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy, 204, 537-547.
- Wang, S., & Stebbins, J. F. (1999). Multiple-quantum magic-angle spinning 17O NMR studies of borate, borosilicate, and boroaluminate glasses. Journal of the American Ceramic Society, 82(6), 1519-1528.
- Wantana, N., Kaewnuam, E., Chanlek, N., Kim, H. J., & Kaewkhao, J. (2023). Influence of Gd3+ on structural and luminescence properties of new Eu³⁺-doped borate glass for photonics material. Radiation Physics and Chemistry, 207, 110812.
- Wright, A. C., Sinclair, R. N., Shaw, J. L., Haworth, R., Bingham, P. A., Forder, S. D., ... & Vedishcheva, N. M. (2017). The environment of Fe³⁺/Fe²⁺ cations in a sodium borosilicate glass. Physics and Chemistry of Glasses-European Journal of Glass Science and Technology Part B, 58(3), 78-91.
- Yano, T., Kunimine, N., Shibata, S., & Yamane, M. (2003). Structural investigation of sodium borate glasses and melts by Raman spectroscopy.: I. Quantitative evaluation of structural units. Journal of Non-Crystalline Solids, 321(3), 137-146.
- Yiannopoulos, Y. D., Chryssikos, G. D., & Kamitsos, E. I. (2001). Structure and properties of alkaline earth borate glasses. Physics and chemistry of glasses, 42(3), 164-172.
- Zhao, H., Zhu, Y., Ye, H., He, Y., Li, H., Sun, Y., ... & Wang, R. (2023). Atomic-scale structure dynamics of nanocrystals revealed by in situ and environmental transmission electron microscopy. Advanced Materials, 35(50), 2206911.
- Zheng, K., & Boccaccini, A. R. (2017). Sol-gel processing of bioactive glass nanoparticles: A review. Advances in Colloid and Interface Science, 249, 363-373.